

LETTERS
TO THE EDITOR

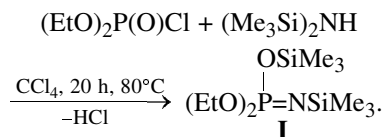
Synthesis
of Diethyl Trimethylsilyl (Trimethylsilylimido)phosphate

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At present two methods of synthesis of dialkyl trimethylsilyl (trimethylsilylimido)phosphates have been described: exhaustive silylation of dialkyl imidophosphate with chlorotrimethylsilane in the presence of bases and the reaction of phosphorus acid chlorides with sodium or lithium bis(trimethylsilyl)amides [1, 2]. According to published data, the reaction of dialkyl(aryl) chlorophosphates with hexamethyldisilazane is accompanied by elimination of trimethylchlorosilane and stops at the stage of formation of dialkyl trimethylsilyl imidophosphates [3, 4]. We found that diethyl trimethylsilyl (trimethylsilylimido)phosphate (**I**) can be prepared by prolonged heating of equimolar amounts of diethyl chlorophosphate and hexamethyldisilazane in carbon tetrachloride. This method is simple in operation and permits one to vary the type of alkyl substituents in the final product.



Diethyl trimethylsilyl (trimethylsilylimido)phosphate (I). A solution of 7.5 g of diethyl chlorophosphate and 7 g of hexamethyldisilazane in 50 ml of

carbon tetrachloride was refluxed for 20 h. The solvent was removed, and the residue was distilled in a vacuum to obtain 5.8 g (45%) of compound **I**, bp 50–53°C (0.06 mm), n_D^{20} 1.4182 [3]. ^{31}P NMR spectrum: δ_P –12.28 ppm. Found, %: P 9.94; Si 18.85. $\text{C}_{10}\text{H}_{28}\text{NOPSi}$. Calculated, %: P 10.43; Si 18.83.

The ^{31}P NMR spectrum was recorded on a Bruker MSL-400 spectrometer (166.93 MHz) against external 85% phosphoric acid.

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